

# Calculating Specific Heat Worksheet

## Answer Key

### SPECIFIC HEAT WORKSHEET

**Specific heat** is the quantity of heat required to raise the temperature of one gram of a substance by one Celsius degree.



$$Q = m c \Delta T$$

where  $Q$  = heat energy,  $m$  = mass, and  $\Delta T$  = change in temp.  
Remember,  $\Delta T = (T_{\text{final}} - T_{\text{initial}})$ .

#### Directions : calculating specific heat

1. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{1086.75}{15.75(175-25)} = 0.46 \text{ J/g}^\circ\text{C}$$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$Q = mC(T_f - T_i) = 10.0\text{g} (0.90\text{J/g}^\circ\text{C})(55-22) = 297 \text{ J}$$

3. Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{67500 \text{ J}}{(1500 \text{ g})(57-32)} = 1.8 \text{ J/g}^\circ\text{C}$$

4. 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_f - T_i) = 100\text{g}(4.184\text{J/g}^\circ\text{C})(37 - 4) = 14000 \text{ J}$$

5. 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the specific heat capacity of mercury.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{455 \text{ J}}{(25\text{g})(155-25)} = 0.14 \text{ J/g}^\circ\text{C}$$

WORKSHEETSGO



## Calculating Specific Heat: Your Ultimate Worksheet Guide

Are you struggling with specific heat calculations? Feeling overwhelmed by formulas and unsure how to apply them to real-world problems? This comprehensive guide provides you with not only a clear explanation of specific heat but also a downloadable calculating specific heat worksheet packed with practice problems to solidify your understanding. We'll break down the concept, explore the formula, and walk you through various examples – all designed to boost your confidence and improve your problem-solving skills. Get ready to master specific heat!

## What is Specific Heat?

Specific heat capacity, often shortened to specific heat, is a fundamental concept in thermodynamics. It represents the amount of heat energy required to raise the temperature of one unit of mass of a substance by one degree Celsius (or one Kelvin). Think of it as a measure of a substance's resistance to temperature change. Some substances heat up quickly with little energy input (low specific heat), while others require significantly more energy to achieve the same temperature increase (high specific heat).

Water, for example, has a relatively high specific heat. This is why oceans regulate Earth's temperature; they absorb and release large amounts of heat with minimal temperature fluctuation. Metals, on the other hand, generally have low specific heat, explaining why they heat up and cool down quickly.

## Understanding the Specific Heat Formula

The core formula used for calculating specific heat is:

$$q = mc\Delta T$$

Where:

$q$  represents the heat energy transferred (usually measured in Joules, J)

$m$  represents the mass of the substance (usually measured in grams, g or kilograms, kg)

$c$  represents the specific heat capacity of the substance (usually measured in J/g°C or J/kg°C)

$\Delta T$  represents the change in temperature (final temperature - initial temperature), measured in °C or K.

Understanding each variable is crucial for accurate calculations. Make sure you are consistent with your units throughout the problem.

## Step-by-Step Guide to Solving Specific Heat Problems

Let's break down the problem-solving process with a practical example:

Problem: How much heat is required to raise the temperature of 50g of water from 20°C to 100°C? The specific heat of water is 4.18 J/g°C.

Step 1: Identify the knowns.

$$m = 50 \text{ g}$$

$$c = 4.18 \text{ J/g}^\circ\text{C}$$

$$\Delta T = 100^\circ\text{C} - 20^\circ\text{C} = 80^\circ\text{C}$$

Step 2: Apply the formula.

$$q = mc\Delta T = (50 \text{ g})(4.18 \text{ J/g}^\circ\text{C})(80^\circ\text{C})$$

Step 3: Calculate the answer.

$$q = 16720 \text{ J}$$

Therefore, 16720 Joules of heat are required.

## Different Units and Conversions

It's essential to be comfortable working with different units. You might encounter problems using kilograms for mass or Kelvin for temperature. Remember to convert units to match the specific heat capacity units provided. For example, if your specific heat is given in J/kg°K, ensure your mass is in kilograms and your temperature change is in Kelvin.

## Advanced Specific Heat Calculations: Phase Changes

The formula  $q = mc\Delta T$  only applies to situations where the substance remains in the same phase (solid, liquid, or gas). When a phase change occurs (e.g., ice melting to water), a different formula involving the heat of fusion or vaporization is necessary. This is a more advanced topic that builds upon the foundation of specific heat calculations.

## Downloadable Calculating Specific Heat Worksheet

To solidify your understanding, we've created a downloadable calculating specific heat worksheet containing a variety of practice problems. These problems range in difficulty, allowing you to build

your skills progressively. This worksheet includes a mix of straightforward calculations and more complex scenarios designed to test your understanding of the concepts and formula applications. [Link to downloadable worksheet would go here]

## Conclusion

Mastering specific heat calculations is vital for understanding thermodynamics and its applications in various scientific fields. By understanding the formula, practicing with example problems, and utilizing our provided calculating specific heat worksheet, you can confidently tackle even the most challenging specific heat problems. Remember to pay close attention to units and be methodical in your approach. Consistent practice is key to success.

## Frequently Asked Questions (FAQs)

1. What happens if I use different units in the specific heat formula? Your answer will be incorrect. Ensure all units are consistent to get the right answer.
2. Can I use Kelvin instead of Celsius for  $\Delta T$ ? Yes, as long as you are consistent with your units throughout the calculation. A change of  $1^{\circ}\text{C}$  is equal to a change of 1K.
3. What if the specific heat isn't given in the problem? You will need to look up the specific heat value for the substance in question in a reference table or textbook.
4. How do I handle phase changes in specific heat calculations? Phase changes require different formulas involving the heat of fusion or vaporization; this is a more advanced topic.
5. Where can I find more practice problems on specific heat? Many textbooks and online resources offer additional practice problems and example calculations. Search for "specific heat problems" online to find numerous resources.

**calculating specific heat worksheet:** *The Development of a Conceptual Framework of Heat*  
John Mackenzie Rogan, 1985

**calculating specific heat worksheet:** University Physics Samuel J. Ling, Jeff Sanny, William Moebs, 2017-12-19 University Physics is designed for the two- or three-semester calculus-based physics course. The text has been developed to meet the scope and sequence of most university physics courses and provides a foundation for a career in mathematics, science, or engineering. The book provides an important opportunity for students to learn the core concepts of physics and understand how those concepts apply to their lives and to the world around them. Due to the comprehensive nature of the material, we are offering the book in three volumes for flexibility and efficiency. Coverage and Scope Our University Physics textbook adheres to the scope and sequence of most two- and three-semester physics courses nationwide. We have worked to make physics interesting and accessible to students while maintaining the mathematical rigor inherent in the

subject. With this objective in mind, the content of this textbook has been developed and arranged to provide a logical progression from fundamental to more advanced concepts, building upon what students have already learned and emphasizing connections between topics and between theory and applications. The goal of each section is to enable students not just to recognize concepts, but to work with them in ways that will be useful in later courses and future careers. The organization and pedagogical features were developed and vetted with feedback from science educators dedicated to the project. VOLUME II Unit 1: Thermodynamics Chapter 1: Temperature and Heat Chapter 2: The Kinetic Theory of Gases Chapter 3: The First Law of Thermodynamics Chapter 4: The Second Law of Thermodynamics Unit 2: Electricity and Magnetism Chapter 5: Electric Charges and Fields Chapter 6: Gauss's Law Chapter 7: Electric Potential Chapter 8: Capacitance Chapter 9: Current and Resistance Chapter 10: Direct-Current Circuits Chapter 11: Magnetic Forces and Fields Chapter 12: Sources of Magnetic Fields Chapter 13: Electromagnetic Induction Chapter 14: Inductance Chapter 15: Alternating-Current Circuits Chapter 16: Electromagnetic Waves

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**calculating specific heat worksheet: Design Manual for Solar Heating of Buildings and Domestic Hot Water** Richard L. Field, 1980

**calculating specific heat worksheet:** *Holt Chemistry* R. Thomas Myers, 2004

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Jasper Green, 2020-07-19 A bullet dropped and a bullet fired from a gun will reach the ground at the same time. Plants get the majority of their mass from the air around them, not the soil beneath them. A smartphone is made from more elements than you. Every day, science teachers get the opportunity to blow students' minds with counter-intuitive, crazy ideas like these. But getting students to understand and remember the science that explains these observations is complex. To help, this book explores how to plan and teach science lessons so that students and teachers are thinking about the right things – that is, the scientific ideas themselves. It introduces you to 13 powerful ideas of science that have the ability to transform how young people see themselves and the world around them. Each chapter tells the story of one powerful idea and how to teach it alongside examples and non-examples from biology, chemistry and physics to show what great science teaching might look like and why. Drawing on evidence about how students learn from cognitive science and research from science education, the book takes you on a journey of how to plan and teach science lessons so students acquire scientific ideas in meaningful ways. Emphasising the important relationship between curriculum, pedagogy and the subject itself, this exciting book will help you teach in a way that captivates and motivates students, allowing them to share in the delight and wonder of the explanatory power of science.

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**calculating specific heat worksheet: Numerical Methods in Thermal Problems** Roland Wynne Lewis, Kenneth Morgan, 1989

**calculating specific heat worksheet: Plant Engineer's Handbook of Formulas, Charts, and Tables** Donald W. Moffat, 1991

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**calculating specific heat worksheet:** *Engineer's Guide to Solar Energy* , 1979

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**calculating specific heat worksheet: Pearson Chemistry 12 New South Wales Skills and Assessment Book** Penny Commons, 2018-10-15 The write-in Skills and Assessment Activity Books focus on working scientifically skills and assessment. They are designed to consolidate concepts learnt in class. Students are also provided with regular opportunities for reflection and self-evaluation throughout the book.

**calculating specific heat worksheet:** *Proceedings of the ASME Turbo Expo ...* , 2005

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the mathematical rigor inherent in the subject. Frequent, strong examples focus on how to approach a problem, how to work with the equations, and how to check and generalize the result.--Open Textbook Library.

**calculating specific heat worksheet:** Refrigeration and Air-conditioning Air-Conditioning and Refrigeration Institute, 1979 Helps prepare readers for the Federally required (EPA) Certification for technicians. Exceptionally comprehensive, authoritative, up-to-date, and well-illustrated in full color. It focuses on accepted and expected industry practices applicable to a wide variety of HVACR jobs. For anyone interested in Basic Refrigeration, Commercial Refrigeration, Residential Air Conditioning, Commercial Air Conditioning. Warm Air Heating, Hydronic Heating, HVAC Control Systems, and Servicing HVAC Systems.

**calculating specific heat worksheet:** The Architects' Journal , 1979-11

**calculating specific heat worksheet:** *Chemical Engineering Design* Gavin Towler, Ray Sinnott, 2012-01-25 Chemical Engineering Design, Second Edition, deals with the application of chemical engineering principles to the design of chemical processes and equipment. Revised throughout, this edition has been specifically developed for the U.S. market. It provides the latest US codes and standards, including API, ASME and ISA design codes and ANSI standards. It contains new discussions of conceptual plant design, flowsheet development, and revamp design; extended coverage of capital cost estimation, process costing, and economics; and new chapters on equipment selection, reactor design, and solids handling processes. A rigorous pedagogy assists learning, with detailed worked examples, end of chapter exercises, plus supporting data, and Excel spreadsheet calculations, plus over 150 Patent References for downloading from the companion website. Extensive instructor resources, including 1170 lecture slides and a fully worked solutions manual are available to adopting instructors. This text is designed for chemical and biochemical engineering students (senior undergraduate year, plus appropriate for capstone design courses where taken, plus graduates) and lecturers/tutors, and professionals in industry (chemical process, biochemical, pharmaceutical, petrochemical sectors). New to this edition: - Revised organization into Part I: Process Design, and Part II: Plant Design. The broad themes of Part I are flowsheet development, economic analysis, safety and environmental impact and optimization. Part II contains chapters on equipment design and selection that can be used as supplements to a lecture course or as essential references for students or practicing engineers working on design projects. - New discussion of conceptual plant design, flowsheet development and revamp design - Significantly increased coverage of capital cost estimation, process costing and economics - New chapters on equipment selection, reactor design and solids handling processes - New sections on fermentation, adsorption, membrane separations, ion exchange and chromatography - Increased coverage of batch processing, food, pharmaceutical and biological processes - All equipment chapters in Part II revised and updated with current information - Updated throughout for latest US codes and standards, including API, ASME and ISA design codes and ANSI standards - Additional worked examples and homework problems - The most complete and up to date coverage of equipment selection - 108 realistic commercial design projects from diverse industries - A rigorous pedagogy assists learning, with detailed worked examples, end of chapter exercises, plus supporting data and Excel spreadsheet calculations plus over 150 Patent References, for downloading from the companion website - Extensive instructor resources: 1170 lecture slides plus fully worked solutions manual available to adopting instructors

**calculating specific heat worksheet:** Holt Science & Technology: Physical Science , 2004

**calculating specific heat worksheet:** CPO Focus on Physical Science CPO Science (Firm), Delta Education (Firm), 2007

**calculating specific heat worksheet:** *Transactions of the American Foundrymen's Society* American Foundrymen's Society, 1982

**calculating specific heat worksheet:** Fundamentals of Heat and Mass Transfer C. P. Kothandaraman, 2006 About the Book: Salient features: A number of Complex problems along with the solutions are provided Objective type questions for self-evaluation and better understanding of



the subject Problems related to the practical aspects of the subject have been worked out Checking the authenticity of dimensional homogeneity in case of all derived equations Validation of numerical solutions by cross checking Plenty of graded exercise problems from simple to complex situations are included Variety of questions have been included for the clear grasping of the basic principles Redrawing of all the figures for more clarity and understanding Radiation shape factor charts and Heisler charts have also been included Essential tables are included The basic topics have been elaborately discussed Presented in a more better and fresher way Contents: An Overview of Heat Transfer Steady State Conduction Conduction with Heat Generation Heat Transfer with Extended Surfaces (FINS) Two Dimensional Steady Heat Conduction Transient Heat Conduction Convection Convective Heat Transfer Practical Correlation Flow Over Surfaces Forced Convection Natural Convection Phase Change Processes Boiling, Condensation, Freezing and Melting Heat Exchangers Thermal Radiation Mass Transfer

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Nicholas P. Chopey, 1994 A compilation of the calculation procedures needed every day on the job by chemical engineers. Tables of Contents: Physical and Chemical Properties; Stoichiometry; Phase Equilibrium; Chemical-Reaction Equilibrium; Reaction Kinetics and Reactor Design; Flow of Fluids and Solids; Heat Transfer; Distillation; Extraction and Leaching; Crystallization; Filtration; Liquid Agitation; Size Reduction; Drying; Evaporation; Environmental Engineering in the Plant. Illustrations. Index.

**calculating specific heat worksheet: The Properties of Gases and Liquids 5E** Bruce E.

Poling, John M. Prausnitz, John P. O'Connell, 2000-11-27 Must-have reference for processes involving liquids, gases, and mixtures Reap the time-saving, mistake-avoiding benefits enjoyed by thousands of chemical and process design engineers, research scientists, and educators. Properties of Gases and Liquids, Fifth Edition, is an all-inclusive, critical survey of the most reliable estimating methods in use today --now completely rewritten and reorganized by Bruce Poling, John Prausnitz, and John O'Connell to reflect every late-breaking development. You get on-the-spot information for estimating both physical and thermodynamic properties in the absence of experimental data with this property data bank of 600+ compound constants. Bridge the gap between theory and practice with this trusted, irreplaceable, and expert-authored expert guide -- the only book that includes a critical analysis of existing methods as well as hands-on practical recommendations. Areas covered include pure component constants; thermodynamic properties of ideal gases, pure components and mixtures; pressure-volume-temperature relationships; vapor pressures and enthalpies of vaporization of pure fluids; fluid phase equilibria in multicomponent systems; viscosity; thermal conductivity; diffusion coefficients; and surface tension.

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**calculating specific heat worksheet: Load Calculation Applications Manual (I-P Edition)**

Jeffrey D. Spitler, 2014-10-01 This manual focuses on the calculation of cooling and heating loads for commercial buildings. The heat balance method (HBM) and radiant time series method (RTSM) (as well as how to implement these methods) are discussed. Heat transfer processes and their analysis, psychrometrics, and heating load calculations are also considered--

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This first volume of Sustainable building design manuals focuses on policy and regulatory mechanisms and serves as a guide to policy-makers and local authorities

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Eckblad, 1978

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### Carson High School (China Grove, NC) Varsity Football - MaxPreps

On Friday, Nov 1, 2024, the Carson Varsity Boys Football team lost their game against South Rowan High School by a score of 14-21.

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Get more information for Jesse C Carson High School in China Grove, NC. See reviews, map, get the address, and find directions.

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JCHS is a North Carolina public high school serving rural southeastern Rowan County, including the small towns of China Grove, Faith, Rockwell, and portions of the City of Salisbury.

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